A Systematic Study of C-Glucoside Trisphosphates as *myo*-Inositol Trisphosphate Receptor Ligands. Synthesis of β -C-Glucoside Trisphosphates Based on the Conformational Restriction Strategy

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 β -C-Glucoside trisphosphates having a C2 side chain (3,7-anhydro-2-deoxy-D-glycero-D-gulo-octitol 1,5.6trisphosphate, 11) and a C3 side chain (4,8-anhydro-2,3-dideoxy-D-glycero-D-gulo-nonanitol 1,6,7trisphosphate, 12) were designed as structurally simplified analogues of a potent D-myo-inositol 1,4,5trisphosphate (IP₃) receptor ligand, adenophostin A. Construction of the β -C-glucosidic structure, which was the key to their synthesis, was achieved by two different methods based on the conformational restriction strategy: (1) radical cyclization with a temporary connecting silicon tether and (2) silane reduction of glyconolactols having an anomeric allyl substituent. Using these methods, the target β -C-glycoside trisphosphates 11 and 12 were successfully synthesized. A structure-activity relationship was established on a series of C-glucoside trisphosphates, including the previously synthesized related compounds, which were a C-glycosidic analogue 3 of adenophostin A, its uracil congener 5, α -C-glucoside trisphosphates 7–9 having a C1, C2, or C3 side chain, and the β -C-glucoside trisphosphates 10–12 having a C1, C2, or C3 side chain. The O-glycosidic linkage of adenophostin A and its analogues proved to be replaced by the chemically and biologically more stable C-glycosidic linkage. The α -C2-glucoside trisphosphate 8 stimulates Ca^{2+} release with a potency similar to that of IP₃ in spite of its simplified structure, indicating a better fit to the receptor than the β -C-glucoside trisphosphates and also the α -congeners having a shorter or longer C1 side chain, which was supported by molecular modeling using the ligand binding domain of the IP_3 receptor.

Introduction

D-*myo*-Inositol 1,4,5-trisphosphate (IP₃, **1**, Figure 1) is produced by phospholipase C-catalyzed hydrolysis of phosphatidylinositol 4,5-bisphosphate, which is stimulated by activation of a diverse array of cell-surface receptors. IP₃ has been shown to mobilize Ca^{2+} from the intracellular stores of most mammalian cells.¹ Many IP₃ analogues have been synthesized as specific ligands of IP₃ receptors. Such analogues may be useful in investigating the mechanisms of IP₃-mediated Ca^{2+} signaling and also as leads for developing novel clinically effective drugs.^{2,3}

In 1993, adenophostin A (2) was isolated from *Penicillium* brevicompactum and shown to be a high-affinity agonist of IP₃ receptors with 10 to ~100 times higher affinity than IP₃.⁴ These results prompted the syntheses of new IP₃ receptor ligands based on the structure of adenophostin A and investigation of the structure—activity relationships of adenophostin A and related compounds.^{5–7} The results showed that the D-glucopyranose structure is a good bioisostere of the *myo*-inositol backbone of IP₃, and that the adenine moiety can be replaced by other aromatic rings: the uracil congener **4**, for example, is almost as potent as adenophostin A in stimulating release of Ca²⁺ from intracellular stores.⁵ Consequently, we became interested in the

C-glucosidic trisphosphates as potential IP₃ receptor ligands, because *C*-glycosides can be biologically stable mimics of the corresponding *O*-glycosides.⁸ Thus, we previously designed and synthesized the *C*-glycosidic analogue **3** of adenophostin A and its uracil congener **5**.^{6d,e} However, their biological effects have not been reported yet.

The three-dimensional positioning of the three phosphate moieties and, in particular, the lone "auxiliary" phosphate group significantly affects the activity of IP₃, adenophostin A, and their derivatives.^{5,6} 2-Hydroxyethyl α -D-glucopyranoside 3,4,2'trisphosphate (6) was originally designed and synthesized as a highly simplified analogue of adenophostin A.7a,b,e Although the O-glycoside trisphosphate **6** is an agonist of the IP₃ receptor, its affinity is more than 10-fold lower than that of IP₃.⁵ We hypothesized that the lower affinity of 6 could be due to its side-chain length, which may be too long to allow the phosphate to achieve an effective position for binding. To test this hypothesis, we designed a series of α -C-glucoside trisphosphates 7-9 having side chains with different lengths, as well as the corresponding β -C-glucoside trisphosphates 10–12. The β -glycosidic structure might be more effective than the α -one, because the lone "auxiliary" phosphate group of IP₃ seems to locate on the β -face of the D-glucose backbone when the inositol and pyranose rings are superimposed. In this strategy, it is essential to employ a series of C-glycosides as stable mimics of the O-glycosides since the corresponding O-glycoside trisphosphates (Figure 2) could not be prepared because of their predictable instability due to the "O-C-O-O-P" or "O-C-O-C-O-

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Figure 1. IP₃ receptor ligands.



Figure 2. Putatively unstable α -O-glucoside trisphosphates.

P" structure. Syntheses of these α -*C*-glycoside trisphosphates **7–9** and the β -C-glucoside trisphosphate **10** with a C1 side chain have been reported.^{6f.g.7c}

In this paper, we describe the synthesis of the β -*C*-glycoside trisphosphate **11** with a C2 side chain and **12** with a C3 side chain, in which the key stereoselective construction of the β -*C*-glucosidic structure was achieved by the conformational restriction strategy using an intramolecular radical cyclization or silane reduction at the anomeric position. The results of a systematic biological evaluation of a series of the new and previously synthesized α - and β -*C*-glucoside trisphosphates are described, and the structure—activity-relationship is discussed together with molecular modeling.

Results and Discussion

 β -Selective Radical Cyclization Based on the Conformational Restriction Strategy. Much attention has been focused on *C*-glycosides because of their importance as stable biologically active mimics for natural *O*-glycosides. Methods for their synthesis have been extensively studied.⁸ However, synthesis of β -*C*-glycosides is considerably more difficult than synthesis of the corresponding α -*C*-glycosides.⁸ Accordingly, effective construction of the β -*C*-glycosidic structure should be the key step in the synthesis of the target compounds in this study.

The use of radical reactions is one of the most efficient methods for constructing C-glycosidic bonds,^{8,9} and therefore we have been working toward the development of stereoselective intramolecular and intermolecular radical C-glycosidation reactions.^{6d-g,9} Stork and co-workers reported an efficient synthesis of a β -C-glucoside via a stereoselective radical cyclization using a 2,3,4-tri-O-benzyl-protected phenyl l-seleno- β -D-glucose derivative having a phenylethynylsilyl group as a radical acceptor tether at the 6-hydroxyl.¹⁰ As shown in Scheme 1a, heating the substrate with Bu₃SnH and AIBN in benzene, followed by treatment with TBAF, gave the 2-phenylvinyl β -Cglucoside in 54% yield. They speculated that the radical cyclization would proceed via a conformationally flipped intermediate (Scheme 1a) in which the tethered hydroxymethyl moiety assumed an axial orientation. With these results in mind, we planned to examine the β -selective introduction of C2 and



C3 units by radical reactions with the corresponding 6-O-vinylsilyl¹¹ and 6-O-allylsilyl¹² substrates.

We previously investigated the introduction of a C2 and a C3 unit stereoselectively at the anomeric α -position of D-glucose via the radical cyclization with a vinylsilyl or allylsilyl group as a temporary connecting tether.^{6f,g} During these studies, we found that the conformational restriction of substrates was effective in realizing highly α -selective reactions.^{6f,g,9} When the substrate with the 2-*O*-allylsilyl group conformationally restricted in the unusual ¹C₄-conformation was used, the desired α -cyclization occurred exclusively (Scheme 1b), while the reaction with the corresponding conformationally unrestricted substrate gave a mixture of the α - and β -products (Scheme 1c).^{6g,i} A similar highly α -selective radical cyclization also occurred with the ¹C₄-restricted substrate with 2-*O*-vinylsilyl group to produce the α -*C*-glycoside having an anomeric C2 chain.^{6f}

Scheme 2



Based on these findings, we decided to apply the conformational restriction strategy to the synthesis of β -C-glycosides via radical cyclization reactions.¹³ Our synthetic plan is shown in Scheme 2. We designed the 6-O-vinylsilyl or O-allylsilyl-Dglucose derivative I, the conformation of which was restricted in a ¹C₄-chair form, as the substrate of radical cyclization reactions. Ab initio calculations suggest that the anomeric radical intermediate preferentially assumes the substratelike ¹C₄-form when the conformations of the precursors of the radical are restricted in an unusual ¹C₄-chair form.^{9a} Therefore, we expected that the radical cyclization using the conformationally ¹C₄restricted substrate I would stereoselectively form the desired β -cyclization product **III**, via the ¹C₄-chair-like anomeric radical intermediate II, in which the *cis*-cyclization would effectively occur without a change of conformation because of the axial orientation of the 6-hydroxymethyl moiety bearing the tether, as shown in Scheme 2. The conformational restriction of the substrates to the desired ¹C₄-form was thought to be possible using significantly bulky silyl protecting groups, as described below. Oxidative cleavage of the Si-C bond of III would give the β -C-glucoside IV, which can be converted into the target trisphosphates 11 or 12.

It is known that introducing a significantly bulky protecting group at the 3,4-*trans*-hydroxyls of pyranoses causes a conformational flip leading to a ${}^{1}C_{4}$ -form, in which the bulky substituents are in axial positions due to mutual steric repulsion.^{9,14,19} Thus, the 3,4-bis-*O*-silylated substrate **13** having a 6-*O*-vinylsilyl group and **14** having a 6-*O*-allylsilyl group (Figure 3), which should be conformationally restricted in the ${}^{1}C_{4}$ -form, were designed for the radical reaction.

The synthesis of the substrates 13 and 14 is summarized in Scheme 3. Starting from D-glucose, phenyl 3,4-bis-O-TBS-6-O-trityl-1-seleno- β -D-glucose (17) was prepared according to the previously reported method.^{6f} Acetylation of the 2-hydroxyl of 17 and subsequent selective removal of the 6-O-trityl group gave 19. Treatment of 19 with vinyldiphenylchlorosilane, DMAP, and Et₃N in toluene at room temperature quantitatively gave the 6-O-vinylsilyl ether 13, the radical reaction substrate. Similarly, the allyldimethylsilyl group was introduced at the 6-hydroxyl with allyldimethylchlorosilane to give the other substrate 14.

We also prepared conformationally unrestricted substrates, i.e., the 6-*O*-allyldimethylsilyl and 6-*O*-vinyldiphenylsilyl ethers of phenyl 2,3,4-tri-*O*-benzyl-l-seleno- β -D-glucose (**15** and **16**,



Figure 3. Conformationally restricted and unrestricted substrates for the radical cyclization reaction.

Scheme 3



Figure 3), to clarify whether the conformational restriction of substrates in the ${}^{1}C_{4}$ -form facilitated the β -selective radical cyclization.

The conformations of the substrates 13-16 were investigated by ¹H NMR. The unrestricted substrates 15 and 16 had large coupling constants (ca. 9 Hz) between the ring protons showing their preference for the usual ⁴C₁-chair-like conformation. On the other hand, the considerably smaller coupling constants between the ring protons in the 3,4-*O*-silyl-protected substrates 13 and 14 indicated that these preferred the flipped ¹C₄-like conformation, as expected.¹³

The radical reactions of the ¹C₄-restricted substrates 13 and 14 as well as the unrestricted substrates 15 and 16 were performed by slow addition of a mixture of Bu₃SnH (1.2 equiv) and AIBN (0.6 equiv) to a heated solution of the substrate in benzene (80 °C). The reaction was carried out first with the ¹C₄-restricted vinylsilyl ether **13** and afforded the desired β -endo-cyclization product **20** in 40% yield. When the ¹C₄restricted allylsilyl ether 14 was subjected to the reaction under the same conditions, the endo-cyclization effectively occurred to give the desired β -product **21** in 72% yield along with the anomeric reduction product 22 in 20% yield. The stereochemistry of the cyclization products 20 and 21 was confirmed by NOE experiments (Figure 4). On the other hand, in the treatment of both of the conformationally unrestricted substrates 15 and 16, under similar Bu₃SnH/AIBN conditions, many spots were observed on TLC, and none of the cyclization products were obtained.



Figure 4. NOE data of the radical cyclization products 20 and 21.



It is worth noting that the reactions of the vinylsilyl and allylsilyl ethers 13 and 14 produced the unusual 8-*endo*- and 9-*endo*-cyclization products 20 and 21, respectively. These results show that the conformational restriction strategy effectively works in the radical cyclization.

Synthesis of the Target Trisphosphate 12. Conversion of the radical reaction products 20 and 21 into the target trisphosphates 11 and 12 was tried (Scheme 4). The 8-membered ring opening via the oxidative Si-C bond fission, without removing the silvl protecting groups at the 3- and 4-hydroxyls, was achieved by treatment with AcOOH/HBr/KBr in DMF,¹⁵ where the β -C-glucosides 23 and 24 were obtained quantitatively and in 61% yield, respectively. Selective silylation of the primary hydroxy on the anomeric chain was next examined. When the hydroxypropyl C-glucoside 24 was treated with TBDPSCl/imidazole in DMF at -40 °C, selective silvlation occurred to give the desired 26 in 64% yield. Treatment of the hydroxyethyl C-glucoside 23 under the same conditions gave the selectively silylated product 25 in only 11% yield. Although we examined selective silvlation of 23 under various conditions, the desired 25 was not obtained selectively.

After benzoylation of the 6-hydroxyl of **26**, the three silyl protecting groups were simultaneously removed to give the triol **28**. The phosphate units were introduced, using the phosphoramidite method with *o*-xylene *N*,*N*-diethylphosphoramidite (XE-PA).¹⁶ Thus, **28** was treated with XEPA and tetrazole in CH₂Cl₂, followed by oxidation with *m*-CPBA to give the desired trisphosphate derivative **29** in 58% yield. Finally, the protecting groups of the phosphates and the hydroxyls were removed by successive hydrogenation and basic hydrolysis to furnish the target **12** quantitatively as a sodium salt, after treatment with ion-exchange resin.

Synthesis of the other target **11** by this route was abandoned because of the very low yield of the anomeric side chain selective silylation of **23**, in which the 6-hydroxyl was preferentially silylated.

Scheme 5



Synthesis of the Target Trisphosphate 11 via Stereoselective Anomeric Silane Reduction with the Conformationally Restricted Substrate. Synthesis of the target 11, however, was accomplished by another route as shown in Scheme 5, via highly stereoselective anomeric deoxygenation based on the conformational restriction strategy.

Kishi reported an efficient synthesis of β -*C*-glycosides, which involves Grignard addition to a glyconolactone to give usually an anomeric mixture of the corresponding lactols bearing an anomeric substituent that is reduced stereoselectively by trialkylsilane under Lewis acidic conditions to the β -*C*-glycoside.¹⁷ However, the stereochemical outcome of the reduction depends on the sugar structure, and the stereoselectivity is not always high.^{17a,d}

We have shown that, in the radical⁹ and also the Lewis acid promoted¹⁸ *C*-glycosidation reactions, the anomeric effect¹⁹ can be controlled by manipulating the substrate conformation, and thus, depending on the conformation of the substrates restricted to the ⁴C₁- or the ¹C₄-form, α - or β -selective reactions can occur highly stereoselectively.^{9,18} Therefore, we expected that in Kishi's silane reduction the stereoselectivity could also be improved by enhancing the kinetic anomeric effect, when conformationally restricted substrates were employed. The conformation of the transition state and the intermediate can be significantly influenced by conformational effects, which stabilize the ground state conformation.^{9,18,20} Accordingly, as shown in Figure 5, in the reduction of substrate **A** conforma-



Figure 5. The working hypothesis for the kinetic anomeric effect dependent α -hydride attack forming the β -*C*-glycosides.

Scheme 6



tionally restricted to a ${}^{4}C_{1}$ -chair form, the transition state would assume the ${}^{4}C_{1}$ -chair-like form **B** due to the conformational restriction of the pyranose backbone, where the anomeric center would be pyramidal. The α -axial attack transition state **B** in the ${}^{4}C_{1}$ -restricted form would be significantly stabilized by the kinetic anomeric effect, i.e., the interaction between the antibonding $\sigma^{*\pm}$ of the newly forming anomeric C–H bond and the orbital of a nonbonded electron pair (n_O) on the ring oxygen, because of their planar arrangement, to produce the β -Cglycoside highly selectively.²¹

Based on this hypothesis, we designed the silane reduction substrate **31** bearing a 3,4-*O*-cyclic-diketal group, the conformation of which would be restricted in the desired ${}^{4}C_{1}$ -form due to the *trans*-decalin-type ring system. 6f,g,9,18,22 The substrate **31** was prepared from D-glucose via an addition of CH₂=CHCH₂-MgBr on the gluconolactone **30**.²³

When the substrate **31** was treated with Et₃SiH and TMSOTf in CH₂Cl₂ at -78 °C, the hydride reduction occurred from the α -side with complete stereoselectivity, as we expected, to give the β -*C*-glycoside **32** in 92% yield. We also examined the reduction of the corresponding conformationally unrestricted tetra-*O*-benzyl substrate **34**, where the stereoselectivity and also the yield clearly decreased (α : β = 9:91, yield 79%),²¹ as shown in Scheme 6. These results demonstrated that the ⁴C₁-conformational restriction effectively improved the stereoselectivity in the Et₃SiH reduction.

Ozonolysis of **32** followed by deprotection with aqueous TFA gave the triol product **33**. Phosphorylation by the phosphoramidite method and subsequent hydrogenation removing the benzyl groups finally afforded the target trisphosphate **11**.

Biological Activity. The ability of a series of *C*-glycosidic compounds, i.e., the *C*-glycosidic adenophostin A analogue **3**, its uracil congener **5**, and the simplified α - and β -*C*-glucoside trisphosphates **7**–**12**, to stimulate opening of the pore of IP₃ receptors was examined. For these analyses, we used recombinant rat type 1 IP₃ receptors stably expressed in chicken B cells that otherwise lack IP₃ receptors and a fluorescent Ca²⁺ indicator trapped within the lumen of the intracellular Ca²⁺ stores to measure the decrease in luminal free [Ca²⁺] that follows activation of IP₃ receptors.^{24–27} The method and its advantages have been recently described.²⁸ The results are summarized in Table 1, and the potencies are presented relative to both IP₃ (**1**) and adenophostin A (**2**).

Maximal concentrations of the analogues 3-5 and 7-12, as well as adenophostin A, examined in this study released the same fraction of the intracellular Ca²⁺ stores (70–80%) as a maximally effective concentration of IP₃ (Table 1), suggesting that each is likely to be a full agonist of the IP₃ receptor. The *C*-glycosidic analogue **3** of adenophostin A was a potent agonist with an EC₅₀ value of 4.0 ± 0.2 nM. It was thus some 6-fold more potent than the natural ligand IP₃ (EC₅₀ = 24.8 ± 2.1 nM) and only 2-fold less potent than adenophostin A (EC₅₀ = 2.1 ± 0.2 nM). The *C*-glycosidic uracil congener **5** was also remarkably active (EC₅₀ = 11.3 ± 2.7 nM), being about 2-fold more potent than IP₃ and only 5-fold less potent than adenophostin A. The effect of replacing the *O*-glycosidic linkage with a *C*-glycosidic linkage was similar for adenophostin A (**2** to **3**)

Table 1. Effects of Compounds on Ca^{2+} Release via Rat Type 1 IP₃ Receptors Expressed in DT40 Cells^{*a*}

	FC	Hill	Ca ²⁺ release		relative potency	
	nM	slope	%	п	1	2
1	24.8 ± 2.1	1.21 ± 0.06	78 ± 2	11	1	0.09 ± 0.01
2	2.1 ± 0.2	1.54 ± 0.13	76 ± 1	12	12.8 ± 1.3	1
3	4.0 ± 0.2	1.56 ± 0.15	81 ± 2	3	6.2 ± 1.2	0.64 ± 0.05
4	4.6 ± 0.4	1.31 ± 0.08	68 ± 2^a	4	6.7 ± 2.4	0.43 ± 0.03
5	11.3 ± 2.7	1.40 ± 0.29	79 ± 3	3	2.4 ± 0.6	0.24 ± 0.05
7	394 ± 39	1.07 ± 0.19	73 ± 1	3	0.11 ± 0.03	0.004 ± 0.001
8	49.2 ± 4.3	1.36 ± 0.15	78 ± 1	5	0.43 ± 0.07	0.04 ± 0.01
9	213 ± 37	1.39 ± 0.27	69 ± 2	5	0.12 ± 0.01	0.009 ± 0.001
10	1080 ± 130	0.98 ± 0.12	70 ± 1	5	0.04 ± 0.01	0.002 ± 0.001
11	2040 ± 430	1.51 ± 0.46	71 ± 3	5	0.014 ± 0.003	0.001 ± 0.0002
12	4430 ± 1090	1.51 ± 0.35	64 ± 5	5	0.007 ± 0.001	0.0006 ± 0.0002

^{*a*} The experiments with **4** were performed separately from the others, although again in parallel with **1** (IP₃) and **2** (adenophostin A); the maximal Ca²⁺ release evoked by **1** and **2** in these experiments was $72 \pm 2\%$ and $69 \pm 3\%$, respectively. It is therefore clear that maximal concentrations of all the ligands listed were as effective as **1** or **2**.

and its uracil congener (4 to 5); in both cases there was a ~ 2 -fold decrease in potency. These results show that, irrespective of the attached nucleobase, changing the glycosidic linkage from -O- to $-CH_2-$ is remarkably well tolerated. Insofar as their interactions with IP₃ receptors are concerned, the three-dimensional structures of the *C*-glycosidic analogues 3 and 5 thus seem to be similar to those of the *O*-glycosides 2 and 4.

The simplified α - and β -*C*-glucoside trisphosphates **7–12** were all less potent than adenophostin A (Table 1). This accords with previous results showing that aromatic rings such as adenine effectively enhance the potency of adenophostin-related compounds.⁵ The activities of these *C*-glucosides were affected by the anomeric α/β -stereochemistry. The β -*C*-glucosides were considerably less potent than the corresponding α -*C*-glucosides. The EC₅₀ values of the β -*C*-glucosides were $\geq 1 \mu$ M, whereas within the α -series, the C2-glucoside **8**, with an EC₅₀ value of 49.2 ± 4.3 nM, was almost comparable to that of IP₃.

The α -C2-glucoside **8** was clearly more potent than either its α -C1 **7** (EC₅₀ = 394 ± 39 nM) or α -C3 **9** (EC₅₀ = 213 ± 37 nM) analogues. Thus, side-chain length markedly affects activity with an optimum chain length of α -C2 in this series. These results suggest that three-dimensional positioning of the "auxiliary" phosphate group in the α -C2-glucoside **8** is suitable for effective receptor binding. It should be noted that **8** is significantly more potent than the simplified *O*-glucoside **6** reported previously, the potency of which is more than 10-fold lower than IP₃,^{5,9b} and thus this systematic study with a series of *C*-glucoside trisphosphates represents a "fine-tuning" of activity related to the auxiliary side chain of this long established prototype analogue.

Modeling Study. The α -*C*-glucoside trisphosphate **8**, which is almost as potent as IP₃, can be used to address the binding mode of adenophostin A and its analogues with a pyranose structure. This is because the simplified structure of **8** makes the conformational analysis easier than with adenophostin A and its analogues reported previously.^{7d}

We calculated the stable three-dimensional structure of the α -C2-glucoside **8** by MacroModel and compared it with that of IP₃. The initial geometries were generated by a conformational search with the Monte Carlo method and were optimized with MMFFs force field. The results are shown in Figure 6. Both compounds have analogous *trans*-vicinal phosphates on the six-membered chair-form ring, and these moieties are well superimposed, as shown in Figure 6c. However, the third phosphate of the α -C2-glucoside **8** is located at a position significantly different from that of IP₃. These results suggest



Figure 6. Stable structures of IP_3 (a) and the α -C2-glucoside 8 (b) calculated by MacroModel with MMFFs force field and their superimposition (c).

that the binding mode of the α -C2-glucoside **8** to the receptor might also be different from that of IP₃.

Potential interactions of the α - and β -*C*-glucoside trisphosphates **7**–**12** with the IP₃ receptor binding site were subsequently explored. Superimposition of **8** and IP₃ indicate that the 3,4-vicinal phosphates of the *C*-glucosides attached directly to the ring may bind to the receptor in the same conformation as the 4,5-vicinal phosphates of IP₃. The *C*-glucoside trisphosphates **7**–**12** were built directly into the IP₃ receptor crystal structure using the bound IP₃ as a template. Energy minimization of each system was performed to identify interactions between the ligand and the receptor.

Figure 7shows the modeling of potential interactions of the α -C2-glucoside 8 (Figure 7a) and the β -C2-glucoside 11 (Figure 7b) with arginine residues, which recognize the phosphates of IP_3 in the binding site of the IP_3 receptor. IP_3 bound in the crystal structure is also shown (Figure 7c). The trans-vicinal phosphate groups of 8 and 11 and the 4- and 5-phosphates of IP_3 have similar conformations; however, it is clear that there are differences between the orientations and interactions of the epimeric side-chain phosphate groups of 8 and 11 and the 1-phosphate group of IP₃. Comparing the conformations of the C-glucosides 8 and 11, it is clear that the most dominant interaction is different between the α - and β -glucosides, in that the phosphate group in the α -glucoside 8 points down toward Arg504 and Arg269, whereas it points toward Arg568 in the β -glucoside 11. IP₃ in the crystal structure has a binding mode similar to that of the compound: Arg568 forms hydrogen bonds with the 1-phosphate of IP₃ and also with the side-chain phosphate of 11. In addition to this, the 1-phosphate group of

IP₃ also interacts with Arg504, Asp566 (not shown), and the peptide backbone of Arg568, mediated by crystal waters. It may be postulated that the difference in activity between the α - and β -glucosides may be related to the interaction of the phosphate group with different arginine residues. Furthermore, from Figure 7a, a suggestion for the higher activity of the C-glucoside 8 may be given in that this is the only glucoside trisphosphate which shows an interaction of the phosphate group with more than one arginine residue, i.e., Arg269 and Arg504. The other two α -C-glucosides 7 and 9, not shown here, show an interaction only with Arg269 and Arg504, respectively. For compound 7 the C1 chain does not seem to be long enough for the phosphate to reach Arg269, and for compound 9 the C3 chain moves the phosphate too far away from Arg504. However, the crystal structure used in this study comprises only the binding domain of the IP₃ receptor. Differences in the observed biological activity might be linked to interactions of ligands with other parts of the receptor.

Conclusion. The synthesis of β -*C*-glucoside trisphosphates **11** and **12** having a C2 side chain or a C3 side chain, designed as structurally simplified analogues of adenophostin A, was achieved. In the synthesis, the key β -*C*-glucosidic structures were effectively constructed based on the conformational restriction strategy using the radical cyclization with a temporary connecting silicon tether or the silane reduction of glyconolactols having an anomeric allyl substituent. We showed that the *O*-glycosidic linkage of adenophostin A and its analogues can be replaced by the chemically and biologically more stable *C*-glucoside trisphosphates related to adenophostin A, the α -C2-glucoside trisphosphate **8** is almost as potent as IP₃ in causing intracellular Ca²⁺ mobilization. This is rationalized in a preliminary fashion using molecular modeling.

Experimental Section

Chemical shifts are reported in ppm downfield from tetramethylsilane (¹H and ¹³C) or H_3PO_4 (³¹P), and coupling constants are given in Hz. All of the ¹H NMR assignments described were in agreement with COSY spectra. Thin-layer chromatography was done on Merck coated plate 60F₂₅₄. Silica gel chromatography was done on Merck silica gel 5715. Reactions were carried out under an argon atmosphere.

Phenyl 2-O-Acetyl-3,4-bis-*O-tert***-butyldimethylsilyl-1-seleno**β-**D-glucopyranoside (19).** A mixture of **17** (1.6 g, 2.0 mmol), Ac₂O (377 μ L, 4.0 mmol) and DMAP (733 mg, 6.0 mmol) in MeCN (20 mL) was stirred at room temperature for 12 h. The mixture was partitioned between AcOEt and H₂O, and the organic layer was washed with brine, dried (Na₂SO₄) and evaporated to give crude **18** (1.7 g, quantitative) as a colorless oil. A mixture of the oil and aqueous TFA (80%, 500 μ L) in CHCl₃ (5 mL) was stirred at room



Figure 7. Calculated binding modes of α -C2-glucoside **8**, β -C2-glucoside **11**, and IP₃: (a) modeling of potential interactions between the α -*C*-glucoside **8** and arginine residues 504 and 269; (b) the β -*C*-glucoside **11** and Arg568; and (c) IP₃ with Arg568 in the ligand binding domain of the IP₃ receptor.

temperature for 1 h and then evaporated. The residue was partitioned between AcOEt and H₂O, and the organic layer was washed with aqueous saturated NaHCO₃ and brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO₂, 5% AcOEt in hexane) to give **19** (1.0 g, 82%) as a white solid: ¹H NMR (400 MHz, CDCl₃) δ 7.58–7.55 (m, 2 H), 7.27–7.24 (m, 3 H), 5.10 (d, 1 H, *J* = 8.2), 4.95 (dd, 1 H, *J* = 5.6, 8.2), 3.79–3.72 (m, 3 H), 3.63–3.58 (m, 3 H), 2.08 (s, 3 H), 0.87 (s, 9 H), 0.86 (s, 9 H), 0.10 (s, 3 H), 0.07 (s, 6 H), 0.06 (s, 3 H); ¹³C NMR (100 MHz, CDCl₃) δ 169.6, 134.5, 133.8, 129.1, 129.0, 127.8, 82.6, 80.5, 74.8, 73.9, 71.0, 62.8, 26.1, 21.6, 18.2, -3.1, -3.3, -3.9, -4.1; LRMS (FAB, positive) *m*/*z* 613 (MNa⁺). Anal. (C₂₆H₄₆O₆SeSi₂): C, H.

Phenyl 2-O-Acetyl-3,4-bis-O-tert-butyldimethylsilyl-6-O-diphenylvinylsilyl-1-seleno- β -D-glucopyranoside (13). A mixture of 19 (680 mg, 1.15 mmol), diphenylvinylchlorosilane (305 µL, 1.38 mmol), Et₃N (231 µL, 1.66 mmol), and DMAP (73 mg, 0.60 mmol) in toluene (23 mL) was stirred at room temperature for 30 min. The mixture was partitioned between AcOEt and H₂O, and the organic layer was washed with brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO2, 0-2% AcOEt in hexane) to give 13 (760 mg, 83%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 7.63–7.09 (m, 15 H, aromatic), 6.48 (dd, 1 H, vinyl-CH₂, J = 15.0, 20.6), 6.27 (dd, 1 H, vinyl-CH, J = 3.8, 15.0), 5.91 (dd, 1 H, vinyl-CH₂, J = 3.8, 20.6), 5.18 (d, 1 H, H-1, J = 8.2), 4.95 (dd, 1 H, H-2, J = 4.2, 8.2), 4.04 (dd, 1 H, H-6a, J = 5.3, 10.5), 3.92 (dd, 1 H, H-5, J = 6.5, 10.5), 3.80-3.77 (m, 3 H, H-3, -H-4, H-6b), 2.07 (s, 3 H, -COCH₃), 0.84 (s, 9 H, -tBu), 0.83 (s, 9 H, -tBu), 0.07 (s, 3 H, -SiCH₃), 0.05 (s, 6 H, -SiCH₃), -0.02 (s, 3 H, -SiCH₃); ¹³C NMR (100 MHz, CDCl₃) δ 169.6, 134.4, 133.8, 129.1, 129.0, 127.8, 82.6, 80.5, 74.8, 73.9, 71.0, 62.8, 26.1, 21.6, 18.2, -3.1, -3.3, -3.2, -4.1; LRMS (FAB, positive) m/z 821 (MNa⁺). Anal. (C₄₀H₅₈O₆SeSi₃): C. H.

Phenyl 2-O-Acetyl-6-O-allyldimethylsilyl-3,4-bis-O-tert-butyldimethylsilyl-1-seleno-β-D-glucopyranoside (14). Compound 14 (495 mg, 90%) was obtained as a colorless oil from 19 (472 mg, 0.80 mmol) as described for the synthesis of 13 with allyldimethylchlorosilane instead of diphenylvinylchlorosilane: ¹H NMR (400 MHz, CDCl₃) δ 7.62-7.57 (m, 2 H, aromatic), 7.34-7.22 (m, 3 H, aromatic), 5.80-5.73 (m, 1 H, allyl-CH), 5.19 (d, 1 H, H-1, J = 7.9), 4.95 (dd, 1 H, H-2, J = 2.0, 7.9), 4.91–4.84 (m, 2 H, allyl-CH₂), 3.90 (dd, 1 H, H-6a, J = 6.2, 9.7), 3.79–3.71 (m, 4 H, G-3, H-4, H-5, H-6b), 2.06 (s, 3 H, -COCH₃), 1.62-1.54 $(m, 2 H, CH_2 = CH - CH_2), 0.91 (s, 9 H, -tBu), 0.90 (s, 9 H, -tBu),$ 0.12 (s, 3 H, -SiCH₃), 0.11 (s, 6 H, -SiCH₃), 0.09 (s, 6 H, -SiCH₃), 0.08 (s, 3 H, -SiCH₃); ¹³C NMR (100 MHz, CDCl₃) δ 169.0, 134.0, 133.8, 133.3, 129.9, 128.8, 127.3, 113.6, 83.3, 80.0, 77.2, 74.4, 74.3, 70.1, 63.2, 26.2, 26.1, 26.0 24.5, 21.4, 18.2, 18.1, -2.3, -2.3, -3.7, -3.9, -4.2, -4.3; LRMS (FAB, positive) m/z711 (MNa⁺). Anal. (C₃₁H₅₆O₆SeSi₃): C, H.

Radical Reaction Product 20. To a refluxing solution of 13 (246 mg, 0.30 mmol) in benzene (60 mL) was added a solution of Bu₃SnH (99 µL, 0.36 mmol) and AIBN (30 mg, 0.18 mmol) in benzene (8.4 mL) dropwise with a syringe pump over 4 h. The resulting mixture was evaporated, and the residue was partitioned between AcOEt and H₂O. The organic layer was washed with brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO₂, 0-2% AcOEt in hexane) to give 20 (77 mg, 40%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 7.63– 7.61 (m, 2 H, aromatic), 7.55–7.53 (m, 2 H, aromatic), 7.41–7.24 (m, 6 H, aromatic), 4.57 (br s, 1 H, 2-CH), 4.32 (dd, 1 H, H-6a, J = 11.7, 11.7), 4.08 (m, 1 H, H-5), 3.97 (dd, 1 H, H-1, J = 4.5, 12.1), 3.79 (br s, 1 H, H-4), 3.67 (dd, 1 H, H-6b, J = 4.4, 11.7), 3.49 (br s, 1 H, H-3), 2.38-2.29 (m, 1 H, H-1'a), 2.07 (s, 3 H, -COCH₃), 1.99–1.89 (m, 1 H, H-1'b), 1.40–1.20 (m, 2 H, H-2'), 0.89 (s, 9 H, -*t*Bu), 0.64 (s, 9 H, -*t*Bu), 0.08 (s, 3 H, -CH₃), 0.04 (s, 3 H, -CH₃), 0.03 (s, 3 H, -CH₃), -0.01 (s, 3 H, -CH₃); LRMS (FAB, positive) m/z 643 (MH⁺). Anal. (C₃₄H₅₄O₆Si₃): C, H.

Radical Reaction Product 21. Compound **21** (115 mg, 72%) was obtained as a colorless oil from **14** (213 mg, 0.30 mmol) as

described for the synthesis of **20**: ¹H NMR (400 MHz, CDCl₃) δ 6.22 (s, 1 H, H-2), 4.31 (dd, 1 H, H-6a, J = 8.5, 8.5), 4.15 (d, 1 H, H-5, J = 8.5), 3.87 (d, 1 H, H-4, J = 12.0), 3.66 (d, 1 H, H-3, J = 12.0), 3.50 (d, 1 H, H-6b, J = 8.5), 2.16–2.10 (m, 2 H, H-1, H-1'a), 2.08 (s, 3 H, –COCH₃), 1.70–1.65 (m, 2 H, H-2'), 1.52–1.47 (m, 1 H, H-1'b), 0.90 (s, 9 H, *-t*Bu), 0.88 (s, 9 H, *-t*Bu), 0.71 (m, 1 H, H-3'a), 0.58 (m, 1 H, H-3'b), 0.14 (s, 3 H, –CH₃), 0.11 (s, 3 H, –CH₃), 0.09 (s, 3 H, –CH₃), 0.06 (s, 3 H, –CH₃), 0.05 (s, 3 H, –CH₃), 0.04 (s, 3 H, –CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 95.8, 77.6, 72.1, 68.3, 62.3, 51.0, 43.2, 26.6, 26.3, 21.4, 18.6, 18.3, 18.2, –0.1, –2.2, –2.8, –3.3, –3.5, –4.8; LRMS (FAB, positive) m/z 533 (MH⁺). Anal. (C₂₅H₅₂O₆Si₃): C, H.

Radical Reactions of 15 and 16. Compounds **15** and **16** were respectively treated under the radical reaction conditions described for the synthesis of **20**. In both cases, the reaction gave many spots on TLC, and none of the desired cyclization product was obtained.

4-O-Acetyl-3,7-anhydro-5,6-bis-O-tert-butyldimethylsilyl-2deoxy-D-glycero-D-gulo-octitol (23). A mixture of 20 (30 mg, 47 μmol), KBr (11 mg, 94 μmol), HBr (30% in AcOH, 3 μL, 12 μmol), and AcOOH (32% in AcOH, 127 µL, 600 µmol) in DMF (1 mL) was stirred at room temperature for 12 h. The mixture was partitioned between AcOEt and H₂O, and the organic layer was washed with aqueous saturated Na₂S₂O₃, aqueous saturated NaH-CO₃, and brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO₂, 25-50% AcOEt in hexane) to give 23 (23 mg, quantitative) as a colorless oil; ¹H NMR (400 MHz, CDCl₃) δ 4.73 (dd, 1 H, J = 6.2, 8.5), 3.80–3.54 (m, 8 H), 3.48 (ddd, 1 H, J = 2.9, 6.2, 6.4), 2.24 (br s, 1 H), 2.06 (s, 3 H), 1.74–1.68 (m, 2 H), 0.89 (s, 9 H), 0.86 (s, 9 H), 0.10 (s, 3 H), 0.09 (s, 6 H), 0.07 (s, 3 H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl3) δ 169.9, 81.40, 76.0, 75.5, 74.6, 72.0, 63.2, 60.1, 34.4, 26.2, 26.1, 21.6, 18.2, 18.2, -2.7, -2.8, -3.8, -3.9; LRMS (FAB, positive) m/z 479 (MH⁺). Anal. (C₂₂H₄₆O₇Si₂): C, H.

5-O-Acetyl-4,8-anhydro-6,7-bis-*O-tert***-butyldimethylsilyl-2,3-dideoxy-D***glycero***-D***gulo***-nonitol (24).** Compound **24** (21 mg, 61%) was obtained as a colorless oil from **21** (35 mg, 0.07 mmol) as described for the synthesis of **23**: ¹H NMR (400 MHz, CDCl₃) δ 5.90 (br s, 1 H), 3.85 (m, 1 H), 3.66–3.62 (m, 7 H), 2.09 (s, 3 H), 1.94 (br s, 1 H), 1.68–1.59 (m, 3 H), 1.23 (m, 2 H), 0.89 (s, 9 H), 0.87 (s, 9 H), 0.08 (s, 6 H), 0.07 (s, 3 H), 0.06 (s, 3 H); ¹³C NMR (100 MHz, CDCl₃) δ 171.1, 93.9, 77.6, 77.2, 72.5, 62.8, 61.8, 26.2, 26.00, 25.7, 21.4, 18.3, 18.1, –3.3, –4.4, –4.5; LRMS (FAB, positive) *m*/*z* 515 (MNa⁺). Anal. (C₂₃H₄₈O₇Si₂): C, H.

4-O-Acetyl-3,7-anhydro-5,6-bis-O-tert-butyldimethylsilyl-1-Otert-butyldiphenylsilyl-2-deoxy-D-glycero-D-gulo-octitol (25). A mixture of 23 (135 mg, 0.28 mmol), TBDPSCl (146 µL, 0.56 mmol), and imidazole (76 mg, 1.1 mmol) in DMF (5 mL) was stirred at -40 °C for 45 min. The mixture was partitioned between AcOEt and H₂O, and the organic layer was washed with brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO₂, 10-20% AcOEt in hexane) to give 25 (23 mg, 11%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 7.68 (d, 4 H, J = 6.8), 7.41–7.35 (m, 6 H), 6.21 (s, 1 H), 3.91–3.77 (m, 4 H), 3.67-3.54 (m, 3 H), 2.13 (m, 2 H), 2.06 (s, 3 H), 2.00 (m, 1 H), 1.78 (m, 1 H), 1.26 (s, 9 H), 0.85 (s, 9 H), 0.84 (s, 9 H), 0.14 (s, 6 H), 0.02 (s, 6 H); ¹³C NMR (100 MHz, CDCl₃) δ 169.1, 135.8, 135.6, 135.5, 135.4, 133.8, 133.3, 129.5, 129.4, 127.6, 127.4, 127.3, 91.7, 77.2, 75.5, 72.7, 69.7, 62.8, 36.6, 27.0, 26.9, 26.1, 26.0, 21.3, 19.5, 18.3, -3.7, -4.5; HRMS (FAB, positive) calcd for C₃₈H₆₄O₇Si₃Na 739.3858 (MNa⁺), found 739.3828.

5-O-Acetyl-4,8-anhydro-6,7-bis-*O-tert***-butyldimethylsilyl-1***-O-tert***-butyldiphenylsilyl-2,3-dideoxy-D***-glycero***-D***-gulo***-nonitol (26).** Compound **26** (11 mg, 64%) was obtained as a colorless oil from **24** (12 mg, 24 μ mol) as described for the synthesis of **25**: ¹H NMR (400 MHz, CDCl₃) δ 7.66 (d, 4 H, J = 6.5), 7.43–7.36 (m, 6 H), 5.92 (br s, 1 H), 3.88 (m, 1 H), 3.68–3.63 (m, 7 H), 2.09 (s, 3 H), 1.95 (br s, 1 H), 1.62–1.44 (m, 4 H), 1.05 (s, 9 H), 0.89 (s, 18 H), 0.10 (s, 9 H), 0.06 (s, 3 H); ¹³C NMR (100 MHz, CDCl₃) δ 169.9, 81.4, 76.0, 75.5, 74.6, 72.0, 63.2, 60.1, 34.4, 26.2, 26.2, 21.6, 18.2, 18.2, -2.7, -2.8, -3.8, -3.9; LRMS (FAB, positive) m/z 753 (MNa⁺). Anal. (C₃₉H₆₆O₇Si₃): C, H.

5-O-Acetyl-4,8-anhydro-9-O-benzoyl-6,7-bis-O-tert-butyldimethylsilyl-1-O-tert-butyldiphenylsilyl-2,3-dideoxy-D-glycero-**D-gulo-nonitol (27).** A mixture of **26** (17 mg, 23 µmol) and BzCl (5.3 µL, 46 µmol) in pyridine (1 mL) was stirred at room temperature for 30 min. The mixture was partitioned between AcOEt and H₂O, and the organic layer was washed with brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO₂, 10% AcOEt in hexane) to give 27 (19 mg, 99%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 8.02-8.00 (m, 2 H), 7.73-7.65 (m, 4 H), 7.56-7.50 (m, 1 H), 7.44-7.35 (m, 8 H), 6.00 (br s, 1 H), 4.58 (d, 1 H, *J* = 10.3), 4.35 (m, 1 H), 3.90 (m, 2 H), 3.76 (dd, 1 H, J = 6.5, 6.5), 3.67 - 3.65 (m, 3 H),2.07 (s, 3 H), 1.99 (m, 1 H), 1.59 (m, 3 H), 1.04 (s, 9 H), 0.90 (s, 18 H), 0.11 (s, 6 H), 0.07 (s, 3 H), 0.06 (s, 3 H); ¹³C NMR (100 MHz, CDCl₃) δ 169.1, 166.1, 135.4, 134.6, 134.4, 133.7, 133.7, 132.8, 130.4, 129.9, 129.6, 129.4, 129.4, 128.7, 128.7, 128.2, 127.6, 127.5, 94.3, 77.6, 77.2, 74.5, 72.6, 63.8, 27.0, 26.7, 26.2, 26.0, 25.9, 25.9, 25.8, 25.8, 25.8, 25.7, 21.3, 19.3, 18.4, 18.1, 18.0, -3.2, -4.3,-4.6; LRMS (FAB, positive) m/z 857 (MNa⁺). Anal. (C₄₆H₇₀O₈-Si₃): C, H.

5-O-Acetyl-4,8-anhydro-9-O-benzoyl-2,3-dideoxy-D-glycero-Dgulo-nonitol (28). A mixture of 27 (19 mg, 23 µmol), TBAF (1 M in THF, 46 μ L, 46 μ mol), and AcOH (13 μ L, 23 μ mol) in THF (1 mL) was stirred at room temperature for 48 h. The mixture was evaporated, and the residue was partitioned between AcOEt and H₂O. The organic layer was washed with brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO₂, 25% AcOEt in hexane) to give **28** (8 mg, 99%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 8.06 (dd, 2 H, J = 7.3, 7.3), 7.58 (dd, 1 H, *J* = 7.3, 7.3), 7.46 (dd, 2 H, *J* = 7.3, 7.3), 4.57 (dd, 1 H, J = 3.2, 12.3), 4.37 (dd, 1 H, J = 1.9, 12.3), 4.12 (dd, 1 H, J = 5.3, 9.4), 3.86 (d, 1 H, J = 9.7), 3.68 - 3.58 (m, 2 H),3.54 (dd, 1 H, J = 9.7, 9.7), 2.11 (s, 5 H), 1.89-1.82 (m, 1 H),1.79-1.70 (m, 1 H), 1.58-1.50 (m, 2 H), 1.37-1.25 (m, 3 H); ¹³C NMR (100 MHz, CDCl₃) δ 169.1, 167.6, 133.4, 129.8, 129.1, 128.4, 94.4, 73.3, 70.1, 67.2, 63.6, 62.1, 43.1, 30.9, 21.3, 20.5, 0.2; LRMS (FAB, positive) *m*/*z* 391 (MNa⁺). Anal. (C₁₈H₂₄O₈): C, H.

4,8-Anhydro-2,3-dideoxy-D-glycero-D-gulo-nonitol 1,6,7-Trisphosphate Derivative 29. A mixture of 28 (28 mg, 76 µmol), XEPA (73 mg, 304 μ mol) and 1*H*-tetrazole (27 mg, 38 μ mol) in CH₂Cl₂ was stirred at 0 °C for 30 min. After addition of H₂O (20 μ L), the mixture was stirred at room temperature for 10 min. The resulting mixture was cooled to -40 °C, and then *m*-CPBA (70 mg, 400 μ mol) was added. The mixture was warmed to room temperature over 30 min and then partitioned between AcOEt and aqueous saturated Na₂S₂O₃. The organic layer was washed with H₂O, aqueous saturated NaHCO₃, and brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO₂, 10% MeOH in CHCl₃) to give 29 (40 mg, 58%) as a yellow foam: ¹H NMR (400 MHz, CDCl₃) δ 8.03–8.01 (m, 2 H), 7.53– 7.11 (m, 15 H), 5.40–4.97 (m, 16 H), 4.71 (dd, 1 H, J = 2.1, 12.3), 4.50 (dd, 1 H, J = 4.1, 12.3), 4.23-4.14 (m, 3 H), 2.59-2.55 (m, 1 H), 2.12 (s, 3 H), 2.12-2.08 (m, 1 H), 1.98-1.91 (m, 1 H), 1.89–1.80 (m, 1 H), 1.65–1.58 (m, 1 H); ³¹P NMR (500 MHz, CDCl₃, H-decoupled) δ -2.18, -2.54, -5.12 (each s); HRMS (FAB, positive) calcd for $C_{42}H_{46}O_{17}P_3$ 915.1948 (MH⁺), found 915.1933.

Sodium 4,8-Anhydro-2,3-dideoxy-D-glycero-D-gulo-nonitol 1,6,7-Trisphosphate (12). A mixture of 29 (40 mg, 44 μ mol) and Pd–C (10%, 52 mg) in MeOH (5 mL) was stirred at room temperature under atmospheric pressure of H₂ for 40 min. The catalysts were filtered off with Celite, and the filtrate was evaporated. A mixture of the residue and NaOMe (28%, 82 μ L, 400 μ mol) in MeOH (5 mL) was stirred at room temperature for 12 h. The mixture was applied to Diaion PK-212 (H⁺-form), and the column was developed with H₂O. The fractions containing 12 (acidic fractions) were evaporated. A solution of the residue was washed with CHCl₃ (three times) and was applied to Diaion WK-100 (Na⁺ form). The column was developed with H₂O, and the fractions containing 12 were evaporated and dried in vacuo to give 12 (sodium salt, 35 mg, quantitative) as a white solid: ¹H NMR (500 MHz, D₂O) δ 4.17 (m, 1 H), 3.90-3.63 (m, 5 H), 3.59-3.49 (m, 3 H), 1.70-1.44 (m, 4 H); ${}^{13}C$ NMR (100 MHz, D_2O) δ 77.5, 75.2, 73.3, 73.0, 71.7, 65.5, 26.8, 21.6; ${}^{31}P$ NMR (500 MHz, D_2O , H-decoupled) δ 2.56, 2.28, 1.72 (each s); HRMS (FAB, negative) calcd for $C_9H_{17}O_{15}$ -Na₃P₃ 526.9473 (M⁻), found 526.9476.

 β -C-Glycoside 32. A mixture of 31 (103 mg, 0.2 mmol), Et₃-SiH (35 μ L, 0.22 mmol), and TMSOTf (40 μ L, 0.22 mmol) in CH₂-Cl₂ (10 mL) was stirred at -78 °C for 1 h, and then Et₃N (100 μ L) was added. The resulting mixture was partitioned between AcOEt and aqueous saturated NaHCO₃, and the organic layer was washed with brine, dried (Na2SO4), and evaporated. The residue was purified by column chromatography (SiO₂, 15% AcOEt in hexane) to give 32 (92 mg, 92%) as a colorless oil: ¹H NMR(400 MHz, CDCl₃) & 7.35-7.25 (m, 10 H), 5.96-5.85 (m, 1 H), 5.11-5.05 (m, 2 H), 4.98 (d, 1 H, J = 10.9), 4.64–4.56 (m, 3 H), 3.89 (dd, 1 H, J = 9.1, 9.7), 3.73 (dd, 1 H, J = 1.8, 11.1), 3.67 (dd, 1 H, J = 5.0, 11.1), 3.66 (dd, 1 H, J = 9.7, 10.2), 3.55 (ddd, 1 H, J =1.8, 5.0, 10.2), 3.42 (dd, 1 H, J = 9.1, 9.1), 3.41 - 3.36 (m, 1 H), 3.30 (s, 3 H), 3.21 (s, 3 H), 2.62-2.56 (m, 1 H), 2.37-2.30 (m, 1 H), 1.36 (s, 3 H), 1.29 (s, 3H); HRMS calcd for C₂₉H₃₈NaO₇ 521.2513 (MNa⁺), found 521.2520.

3,7-Anhydro-4,8-di-O-benzyl-2-deoxy-D-glycero-D-gulo-octitol (33). Ozone-containing oxygen was bubbled into a solution of 32 (100 mg, 0.20 mmol) in MeOH (2 mL) at -20 °C until 32 disappeared on TLC (ca. 20 min). After addition of NaBH₄ (23 mg, 0.60 mmol) at the same temperature, the resulting solution was stirred at room temperature for 1 h. The mixture was evaporated, and the residue was purified by column chromatography (SiO₂, 30-50% AcOEt in hexane) to give the crude ozonolysis product (47 mg) as an oil. A mixture of the oil and aqueous TFA (80%, 200 µL) in CHCl₃ (2 mL) was stirred at 0 °C for 12 h, and then aqueous saturated NaHCO₃ (2 mL) was added. The resulting mixture was partitioned between CHCl₃ and the aqueous saturated NaHCO₃, and the organic layer was washed with brine, dried (Na₂SO₄), and evaporated. The residue was purified by column chromatography (SiO₂, 10% MeOH in CHCl₃) to give 33 (59 mg, 60%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 7.36–7.26 (m, 10 H), 4.83 (d, 1 H, J = 10.8), 4.71 (d, 1 H, J = 10.8), 4.57–4.55 (m, 2 H), 4.17 (br, 1 H), 3.86-3.78-3.63 (m, 9 H), 3.26-3.21 (m, 2 H), 1.98-1.82 (m, 2 H); LRMS (FAB, positive) m/z 411 (MNa⁺). Anal. (C₂₂H₂₈O₆): C, H.

Sodium 3,7-Anhydro-2-deoxy-D-glycero-D-gulo-octitol 1,5,6-Trisphosphate (11). A mixture of 33 (7.6 mg, 20 µmol), XEPA (19 mg, 80 μ mol), and 1*H*-tetrazole (7.0 mg, 10 μ mol) in CH₂Cl₂ (1 mL) was stirred at 0 °C for 60 min. After addition of H₂O (20 μ L), the mixture was stirred at room temperature for 10 min. The resulting mixture was cooled to -40 °C, and then m-CPBA (40 mg, 200 μ mol) was added. The mixture was warmed to room temperature over 30 min. The reaction mixture was partitioned between AcOEt and aqueous saturated Na₂S₂O₃, and the organic layer was washed with H₂O, aqueous saturated NaHCO₃, and brine, dried (Na₂SO₄), and evaporated. The residue was purified by preparative TLC (SiO₂, 10% MeOH in CHCl₃) to give the phosphorylation product as a foam (13 mg). A mixture of a foam and Pd-C (10%, 17 mg) in MeOH (2 mL) was stirred at room temperature under atmospheric pressure of H₂ for 4 h. After filtration of the catalysts with Celite, the filtrate was evaporated. From the residue, compound 11 (sodium salt, 8 mg, 70% from 33) was obtained as a white solid, as described for the purification of **12**: ¹H NMR (500 MHz, D₂O) δ 3. (dd, 1 H, J = 8.4, 17.0), 3.80 (dd, 1 H, J = 8.4, 10.0), 3.78 (dd, 1 H, J = 8.4, 8.8), 3.66–3.12 (m, 6 H), 2.02 (m, 1 H), 1.55 (m, 1 H); ¹³C NMR (100 MHz, D₂O) δ 81.4, 74.4, 73.3, 70.4, 69.7, 62.1, 33.8, 21.1; ³¹P NMR (500 MHz, D_2O , H-decoupled) δ 3.11, 2.67, 2.36 (each s); HRMS (FAB) calcd $C_8H_{14}O_{15}Na_6P_3$ 580.8932 (MH⁺), found 580.8914.

Modeling Studies. Initial investigations into binding modes of the *C*-glucoside trisphosphates at the IP₃ receptor (PDB code 1N4K) were performed using a docking program (GOLD version 2.2). Docks carried out both in the presence and in the absence of conserved crystal waters gave poor results as the glucose ring was in a very different orientation to the ring of the crystal structure

IP₃ ligand. As an alternative method, simulations of the ligand binding mode for the C-glucoside trisphosphates 7-12 were carried out using Molecular Operating Environment (MOE 2004.03). The basic assumption behind this study was that the two phosphate groups attached directly to the ring bind in the same conformation as the 4- and 5-position phosphate groups of IP₃. Therefore, it was logical to use the crystal structure conformation of IP3 bound to the IP₃ receptor as a starting point for building these molecules. Heavy atoms of the IP3 binding site (PDB code 1N4K) and associated water molecules were fixed, and hydrogen atoms were added. The system was energy-minimized using the MMFF94 force field in MOE to remove any clashes between the newly added hydrogen atoms. A series of modifications were implemented to transform the IP₃ structure into the six *C*-glycoside trisphosphates. These can be summarized as follows: (i) replacement of carbon at position 2 with oxygen; (ii) replacement of OH attached at the 3-position with CH₂OH; (iii) addition of appropriate $(CH_2)_n OPO_3^{2-1}$. where n = 1, 2, or 3 in the correct position to form the three α -Cglucoside and three β -C-glucoside polyphosphates 7–12. The system was energy-minimized following each transformation. Once the desired structures had been built, observation of the resulting conformations suggested possible interactions with three arginine residues across the α - and β -C-glucosidic compounds, namely, Arg269 and Arg504 (for the α -structures) and Arg568 and Arg269 (for the β -structures). Therefore, a final minimization step was simulated by allowing these three residues to move. Interactions of 8, 11, and IP₃ with these active site arginines are shown in Figure 7a-c.

Biological Assay. The ability of compounds to stimulate IP_3 receptors was measured using a low-affinity Ca^{2+} indicator trapped within the intracellular stores of chicken DT 40 cells expressing only recombinant rat type 1 IP_3 receptors as previously.²⁸

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Supporting Information Available: Analytical data of compounds 13, 14, 19–21, 23, 24, 26–28, and 33. This material is available free of charge via the Internet at http://pubs.acs.org.

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